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Structure-Activity Studies of *N*-Cyano-3-pyridinecarboxamides and their Amide and Thioamide Congeners

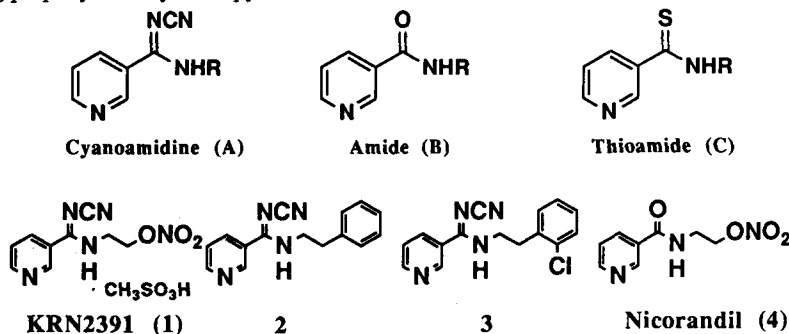
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Abstract. The vasorelaxant and cardiac effects of *N*-cyano-3-pyridinecarboxamide **A** were compared with those of their amide (**B**) or thioamide (**C**) congener. Cyanoamide was the most potent skeleton in comparison with amide or thioamide, and the activities of these three compounds are more selective for vascular smooth muscle than myocardium.

In the course of our studies to explore a new structural class of potassium channel openers, we designed heteroaromatic cyanoamide as a common structural feature based on the structures of nicorandil (**4**)¹ and pinacidil² and found that *N*-cyano-*N'*-substituted 3-pyridinecarboxamide **A** had a noticeable character.³ We first reported that *N*-cyano-*N'*-(2-nitroxyethyl)-3-pyridinecarboxamide methanesulfonate (**1**; KRN2391) was a novel potassium channel opener.^{3a} The *N*-substituent of **1**, *i.e.*, 2-nitroxyethyl, was thought to play an important role not only for a nitrate action but also for a potassium channel opening action. We next investigated to optimize the vasoactivity of cyanoamide without a nitroxyl group, and found that a phenethyl or a 2-(2-chlorophenyl)ethyl group was effective as a *N*-substituent.^{3b} *N*-Cyano-*N'*-phenethyl-3-pyridinecarboxamide (**2**) and *N*-[2-(2-chlorophenyl)ethyl]-*N'*-cyano-3-pyridinecarboxamide (**3**) showed good vasodilating abilities based on a potassium channel opening action. These results revealed that a nitroxyl moiety was not essential to show potassium channel opening property of *N*-cyano-3-pyridinecarboxamides.



In this paper, we focused our attention to the structure-activity relationship of *N*-cyano-*N'*-substituted 3-pyridinecarboxamide **A** versus their amide (**B**) and thioamide (**C**) congeners on

the vasodilating, chronotropic, and inotropic effects. *N*-Substituents were fixed to 2-nitroxyethyl, phenethyl, and 2-(2-chlorophenyl)ethyl groups, respectively. We first compared vasodilating activities of these compounds with their antagonism of 25 mM KCl-induced contraction in isolated rat aorta. Next, the chronotropic and inotropic effects were also examined using isolated guinea-pig right and left atria, respectively.

Chart 1

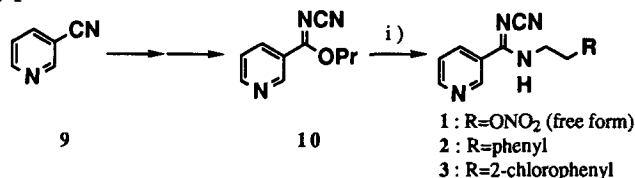


Chart 2

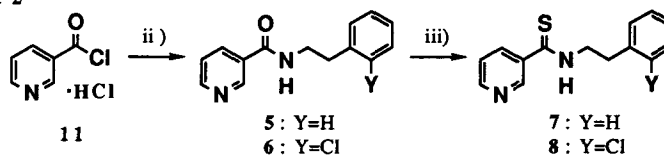
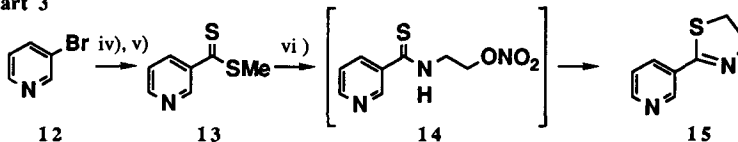


Chart 3



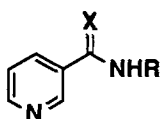
i) H₂N(CH₂)₂R, MeOH; ii) 2-(2-Y-phenyl)ethylamine, Et₃N, CH₃CN; iii) Lawesson's reagent, PhMe, reflux; iv) BuLi / ether-hexane, -76 – -70 °C; v) CS₂, -76 °C – r.t. then MeI; vi) H₂N(CH₂)₂ONO₂ · HCl, NaOMe / MeOH

Cyanoamidines 1, 2, and 3 were prepared from propyl *N*-cyano-3-pyridinecarboximidate (10) by treatment with requisite amines according to our reported procedure (Chart 1).³ Amides 5 and 6 were synthesized by the reaction of nicotinoyl chloride hydrochloride (11) and corresponding amines (Compound 5: mp 77 °C, 60%; compound 6: syrup, 73%). Treatment of amides 5 and 6 with Lawesson's reagent in refluxing toluene afforded thioamides 7 and 8, respectively (Compound 7: mp 106–109 °C, 42%; compound 8: mp 90 °C, 49%) (Chart 2).⁴ Attempts to make 14, the thioamide analogue of nicorandil (4), were failed. We have previously reported that 4 could not be transformed to the corresponding thioamide 14 by treatment with Lawesson's reagent.^{3a} In this study, we tried to prepare 14 *via* dithioester 13 (Chart 3). Lithiation of 3-bromopyridine (12) with butyl lithium and subsequent addition of carbon disulfide, followed by methyl iodide afforded 13.⁵ Reaction of 13 with 2-nitroxyethylamine could not afford the desired thioamide 14 but 3-(2-thiazolin-2-yl) pyridine (15) as a sole product. This result suggested that 13 was first converted to the desired 14, which spontaneously cyclized to the thiazoline 15 because of its unstability.

The vasodilating effect of compounds 1-8 was studied in isolated rat aorta contracted by 25 mM KCl (see Table).⁶ Compounds 1-8 produced a concentration-dependent relaxation. Table shows pD_2 values ($-\log EC_{50}$) calculated by using a concentration-response curve of each compound.⁶ Comparing the vasodilating activity of cyanoamidine, amide, and thioamide with the same *N*'-substituents, the order of relaxant potency was cyanoamidine > thioamide > amide. 2-(2-Chlorophenyl)ethyl substituent was more potent than phenethyl counterpart. These results suggested that two parts in the molecule, *i.e.* a group directly attached to pyridine ring and *N*-substituent, play an important role in the vasodilating potency of the 3-pyridine compounds. The importance of *N*-substituent for the vasodilating potency was reported in the previous papers.³

The chronotropic and inotropic effects of compounds 1-8 were also examined using isolated guinea-pig right and left atria, respectively (see Table).⁷ Cyanoamidines and thioamides produced negative chronotropic and inotropic effects. In contrast, amides had relatively weak action on both atria. In cardiac function, cyanoamidine analogues showed the most potent action as compared with amides or thioamides with the same *N*-substituents. Furthermore, among the three types of 3-pyridine derivatives, *i.e.*, cyanoamidine, amide, and thioamide, the compounds with 2-(2-chlorophenyl)ethyl substituent produced more potent action. These structure-activity relationships in myocardium were similar to those in vascular smooth muscle. However, the concentration of compounds 1-8 to elicit cardiac effects was higher than that to induce their vasorelaxant effects. Therefore, it is considered that compounds 1-8 are more selective for vascular smooth muscle than myocardium.

Table. The pD_2 Values for the Negative Chronotropic (Right Atria), Negative Inotropic (Left Atria), and Vasorelaxing (Aorta) Effects



Comp.			pD_2		
No.	X	R	right atria	left atria	aorta
1 ^{a)}	NCN	2-nitroxyethyl	$4.42 \pm 0.20^{***}$	$4.84 \pm 0.07^{***}$	6.95 ± 0.11
2	NCN	phenethyl	$3.85 \pm 0.22^{***}$	$4.60 \pm 0.06^{***}$	6.40 ± 0.12
3	NCN	2-(2-chlorophenyl)ethyl	$5.32 \pm 0.04^{***}$	$5.51 \pm 0.41^{***}$	7.57 ± 0.16
4	O	2-nitroxyethyl	<3.5	<3.5	5.62 ± 0.20
5	O	phenethyl	<3.5	<3.5	4.09 ± 0.03
6	O	2-(2-chlorophenyl)ethyl	<3.5	$3.70 \pm 0.09^{***}$	4.83 ± 0.04
7	S	phenethyl	$4.10 \pm 0.11^{**}$	$4.39 \pm 0.19^*$	4.95 ± 0.03
8	S	2-(2-chlorophenyl)ethyl	$4.54 \pm 0.20^{**}$	<4.5	5.64 ± 0.04

* $p < 0.05$, ** $P < 0.01$, *** $P < 0.001$: Significant difference from the values obtained in aorta. See references 4 and 5 for experimental details. a) Methanesulfonate.

References and Footnotes

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